Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

Sarah A. Barnett, ${ }^{\text {a* }}$ Andrea Johnston, ${ }^{\text {b }}$ Alastair J. Florence ${ }^{\text {b }}$ and Alan R. Kennedy ${ }^{\text {c }}$

${ }^{\text {a }}$ Department of Theoretical and Computational Chemistry, University College London, 20 Gordon Street, London WC1H 0AJ, England,
${ }^{\mathbf{b}}$ Department of Pharmaceutical Sciences, University of Strathclyde, 27 Taylor Street, Glasgow G4 0NR, Scotland, and ${ }^{c}$ Department of Pure and Applied Chemistry, University of Strathclyde, 295 Cathedral Street, Glasgow G1 1XL, Scotland

Correspondence e-mail:
sarah.barnett@ucl.ac.uk

## Key indicators

Single-crystal X-ray study
$T=123 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.038$
$w R$ factor $=0.080$
Data-to-parameter ratio $=14.7$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
(C) 2005 International Union of Crystallography Printed in Great Britain - all rights reserved

## 3,4-Dichloro-1-nitrobenzene-aniline (2/1)

The solvate structure of 3,4-dichloro-1-nitrobenzene with aniline, $2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{Cl}_{2} \mathrm{NO}_{2} \cdot \mathrm{C}_{6} \mathrm{H}_{7} \mathrm{~N}$, is reported. Ribbons of 3,4dichloronitrobenzene, formed by $\mathrm{Cl} \cdots \mathrm{Cl}$ and $\mathrm{N}-\mathrm{O} \cdots \mathrm{Cl}$ interactions, are linked together via $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds with aniline into an undulating two-dimensional sheet.

## Comment

The title compound, (I), was produced during an automated parallel crystallization polymorph screen on 3,4-dichloronitrobenzene ( $3,4-\mathrm{DCNB}$ ). The sample was identified as a novel form using multi-sample X-ray powder diffraction analysis of all recrystallized samples (Florence et al., 2003). Subsequent manual recrystallization from a saturated aniline solution by slow evaporation at 298 K yielded samples suitable for single-crystal X-ray analysis. The title solvate, (I), crystallizes in the space group $P 2_{1} / n$ with two molecules of $3,4-$ DCNB and one molecule of aniline in the asymmetric unit (Fig. 1).


The crystal structure of (I) is characterized by ribbons of $3,4-\mathrm{DCNB}$, which are linked by aniline molecules to form a


The asymmetric unit of (I), showing the numbering scheme used. Displacement ellipsoids are drawn at the $50 \%$ probability level.

Received 22 June 2005 Accepted 24 June 2005 Online 30 June 2005
$\qquad$


Figure 2
The two-dimensional network formed by (I), showing the intermolecular interactions involved as dashed lines (3,4-DCNB molecule 1: green; 3,4DCNB molecule 2: blue; aniline: red).


Figure 3
Packing diagram viewed perpendicular to the sheets, illustrating the out-of-plane aniline molecules and the stacking arrangement of the sheets. Intermolecular interactions are shown as dashed lines.
continuous sheet (Fig. 2). Molecules of type 1 (C1-C6) form a zigzag chain via $\mathrm{Cl} \cdots \mathrm{Cl}$ interactions $\left[\mathrm{Cl} 1 \cdots \mathrm{Cl} 2^{\mathrm{i}}=3.399\right.$ (1) $\AA$ and $\mathrm{C} 3-\mathrm{Cl} 1 \cdots \mathrm{Cl} 2^{\mathrm{i}}=149.4(1)^{\circ}$; symmetry code: (i) $\frac{3}{2}-x, \frac{1}{2}+$ $\left.y, \frac{3}{2}-z\right]$. These molecules are then involved in a second contact with molecules of type $2(\mathrm{C} 7-\mathrm{C} 12)$ via $\mathrm{N}-\mathrm{O} \cdots \mathrm{Cl}$ interactions $\left[\mathrm{O} 2 \cdots \mathrm{Cl} 4^{\mathrm{ii}}=3.056\right.$ (2) $\AA$ and $\mathrm{N} 1-\mathrm{O} 2 \cdots \mathrm{Cl} 4^{\mathrm{ii}}=$ $140.1(1)^{\circ}$; symmetry code: (ii) $1-x, 1-y, 1-z$ ], thus forming $3,4-\mathrm{DCNB}$ ribbons running parallel to the $b$ axis. The aniline solvent molecules, which lie in a perpendicular plane, link these ribbons into an undulating sheet through two $\mathrm{N}-$ $\mathrm{H} \cdots \mathrm{O}$ interactions $\left[\mathrm{N} 3 \cdots \mathrm{O} 1^{\mathrm{iii}}=2.52(2) \AA\right.$ and $\mathrm{N} 3-$ $\mathrm{H} 1 \mathrm{~N} \cdots \mathrm{O}{ }^{\mathrm{iii}}=157(2)^{\circ}$, and $\mathrm{N} 3 \cdots \mathrm{O} 3^{\mathrm{i}}=2.64(2) \AA$ and $\mathrm{N} 3-$ $\mathrm{H} 2 \mathrm{~N} \cdots \mathrm{O}^{\mathrm{i}}=147(2)^{\circ}$; symmetry code: (iii) $\left.2-x, 1-y, 1-z\right]$. These sheets form an interdigitated $A B A B$ stack parallel to the $a$ axis (Fig. 3).

## Experimental

A single-crystal sample of the title compound was recrystallized from aniline solution by slow evaporation at ca 293 K .

## Crystal data

$2 \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{Cl}_{2} \mathrm{NO}_{2} \cdot \mathrm{C}_{6} \mathrm{H}_{7} \mathrm{~N}$
$M_{r}=477.11$
Monoclinic, $P 2_{1} / n$
$a=6.9774(2) \AA$
$b=10.1668(3) \AA$
$c=27.6762(7) \AA$
$\beta=96.495(2)^{\circ}$
$V=1950.69(9) \AA^{3}$
$Z=4$
$D_{x}=1.625 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 4734
reflections
$\theta=1.0-27.9^{\circ}$
$\mu=0.64 \mathrm{~mm}^{-1}$
$T=123$ (2) K
Triangle, orange
$0.45 \times 0.30 \times 0.15 \mathrm{~mm}$
Data collection
Nonius KappaCCD diffractometer $\omega$ and $\varphi$ scans
Absorption correction: none
22354 measured reflections
4629 independent reflections
3314 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.038$
$w R\left(F^{2}\right)=0.081$
$S=1.04$
4629 reflections
314 parameters
All H -atom parameters refined

$$
\begin{aligned}
& R_{\text {int }}=0.058 \\
& \theta_{\max }=27.9^{\circ} \\
& h=-9 \rightarrow 9 \\
& k=-13 \rightarrow 13 \\
& l=-36 \rightarrow 36
\end{aligned}
$$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0275 P)^{2}\right. \\
& \quad+0.7246 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.002 \\
& \Delta \rho_{\max }=0.28 \mathrm{e}^{2} \AA^{-3} \\
& \Delta \rho_{\min }=-0.32 \mathrm{e}^{-3}
\end{aligned}
$$

$\mathrm{C}-\mathrm{H}$ distances are in the range $0.09(2)-1.00(2) \AA$, and $\mathrm{N}-\mathrm{H}$ distances are 0.86 (2) and 0.87 (2) $\AA$.

Data collection: COLLECT (Hooft, 1998) and DENZO (Otwinowski \& Minor, 1997) and COLLECT; cell refinement: DENZO; data reduction: DENZO; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2000) and OLEX (Dolomanov et al., 2003); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2003).

The authors acknowledge the Research Councils UK Basic Technology Programme for supporting 'Control and Prediction of the Organic Solid State' (www.cposs.org.uk).

## References

Bruker (2000). SHELXTL. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.
Dolomanov, O. V., Blake, A. J., Champness, N. R. \& Schröder, M. (2003). J. Appl. Cryst. 36, 1283-1284.
Florence, A. J., Baumgartner, B., Weston, C., Shankland, N., Kennedy, A. R., Shankland, K. \& David, W. I. F. (2003). J. Pharm. Sci. 92, 1930-1938.
Hooft, R. (1998). COLLECT. Nonius BV, Delft, The Netherlands.
Otwinowski, Z. \& Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr \& R. M. Sweet, pp. 307-326. New York: Academic Press.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.

